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UTILITY **PATENT APPLICATION TRANSMITTAL**

Attorney Docket No. First Inventor Title

Express Mail Label No. (Only for new nonprovisional applications under 37 CFR 1.53(b)) Assistant Commissioner for Patents **APPLICATION ELEMENTS** ADDRESS TO: Box Patent Application Washington, DC 20231 See MPEP chapter 600 concerning utility patent application contents Fee Transmittal Form (e.g., PTO/SB/17) CD-ROM or CD-R in duplicate, large table or Computer Program (Appendix) (Submit an original and a duplicate for fee processing) Applicant claims small entity status. 8. Nucleotide and/or Amino Acid Sequence Submission (if applicable, all necessary) See 37 CFR 1.27. (preferred arrangement set forth below) Computer Readable Form (CRF) b. Specification Sequence Listing on: - Descriptive title of the invention Cross Reference to Related Applications i. ☐ CD-ROM or CD-R (2 copies); or - Statement Regarding Fed sponsored R & D - Reference to sequence listing, a table, ii.□ paper or a computer program listing appendix Statements verifying identity of above copies - Background of the Invention - Brief Summary of the Invention ACCOMPANYING APPLICATION PARTS - Brief Description of the Drawings (if filed) Assignment Papers (cover sheet & document(s)) - Detailed Description - Claim(s) 37 CFR 3.73(b) Statement Power of Attorney - Abstract of the Disclosure (when there is an assignee) English Translation Document (if applicable) Drawing(s) (35 U.S.C. 113) [Total Sheets [] Copies of IDS Information Disclosure [Total Pages [____] Citations 5. Oath or Declaration Statement (IDS)/PTO-1449 **Preliminary Amendment** Newly executed (original or copy) Copy from a prior application (37 CFR 1.63 (d)) (for continuation/divisional with Box 17 completed) Return Receipt Postcard (MPEP 503) b. 14 (Should be specifically itemized) Certified Copy of Priority Document(s) (if foreign priority is claimed) **DELETION OF INVENTOR(S)** 15. Signed statement attached deleting inventor(s) named in the prior application, see 37 CFR 16. 1.63(d)(2) and 1 33(b). Application Data Sheet. See 37 CFR 1.76 17. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment, or in an Application Data Sheet under 37 CFR 1.76: of prior application No.: Continuation-in-part (CIP) Divisional Continuation Group / Art Unit Prior application information: Examiner For CONTINUATION OR DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 5b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation can only be relied upon when a portion has been inadvertently omitted from the submitted application parts. 18. CORRESPONDENCE ADDRESS Correspondence address below Customer Number or Bar Code Label (Insen Customer No. or Attach bar code label here) Name Address Zip Code State City Fax Telephone Country Registration No. (Attorney/Agent) Name (Print/Type)

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November 4, 2000

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Christopher E. Blank

Reg. No. 31,237

BGB Legal Services

4 Bicentennial Square

Suite 2b

Concord, New Hampshire 03301

Ph 603 227 5248

Fx 603 227 5263

Improved Ophthalmic and Contact Lens Wetting Solutions

Inventors

Francis Xavier Smith

John Randall Tracy

Background

The present invention relates to novel ophthalmic solutions that contain a ethoxylated glyceride as an additive to improve the wettability and to decrease the degree of protein and polymeric preservative binding to contact lens surfaces. These compositions may also comprise other agents in contact lens and ophthalmic solutions such as buffers, tonicity agents, wetting agents, enzymes, hydrogen peroxide, demulcents, thickeners, sequestering agents (chelating agents), surface active agents and preservative agents. The ethoxylated glycerides are particularly useful in contact lens treatment solutions, contact lens wetting solutions, solutions used to store contact lenses and solutions used to clean or rinse contact lenses. It has been found that surprisingly the addition of ethoxylated glycerides improve the comfort of lenses treated with such solution and that this increased comfort is surprisingly long-lasting in its effect. The ethoxylated glycerides may be mono-, di- or triglycerides and include

The solutions of the present invention are made by one of two methods. First the ethoxylated glyceride may be melted and added to an aqueous solution which includes the other agents to be used in the desired formulation, or the additional agents may be added prior to the addition of the melted ethoxylated glyceride. Second, the ethoxylated glyceride may be dissolved in an alcohol base and this liquid mixture, added to the aqueous base. Ethoxylated glycerides are commercially available from numerous commercial sourcesand include Polyoxyl 40 hydrogenated castor oil (Cremophor RH 40), polyoxyl 60 hydrogenated castor oil (Cremophor RH 60), PEG-30 Castor Oil (Incrocas 30), PEG-35 Castor Oil (Cremophor EL, Incrocas 35), or PEG-40 Castor Oil (Cremophor EL, Incrocas), Cremophor EL ®, Emulphor EL ®, glycerol polyethyleneglycol riciinoleate, gycerol polyethyleneglycol oxystearate, polyethoxylated hydrogenated castor oil, or polyethoxylated vegetable oil. The ethoxylated glycerides useful in the present invention may include surfactants sold as PEG-6 Caprylic/Capric Glycerides PEG-8 Caprylic/Capric Glycerides; PEG-2 Castor Oil; PEG-3 Castor Oil; PEG-4 Castor Oil; PEG-5 Castor Oil; PEG-8 Castor Oil; PEG-9 Castor Oil; PEG-10 Castor Oil; PEG-11 Castor Oil; PEG-15 Castor Oil; PEG-20 Castor Oil; PEG-25 Castor Oil; PEG-30 Castor Oil; PEG-33 Castor Oil; PEG-35 Castor Oil; PEG-36 Castor Oil; PEG-40 Castor Oil; PEG-50 Castor Oil; PEG-54 Castor Oil; PEG-55 Castor Oil; PEG-60 Castor Oil; PEG-100 Castor Oil; PEG-200 Castor Oil; PEG-18 Castor Oil Dioleate; PEG-60 Corn Glycerides; PEG-20 Evening Primrose Glycerides;

PEG-60 Evening Primrose Glycerides; PEG-7 Glyceryl Cocoate; PEG-30 Glyceryl Cocoate; PEG-78 Glyceryl Cocoate; PEG-80 Glyceryl Cocoate; PEG-12 Glyceryl Dioleate; PEG-15 Glyceryl Isostearate; PEG-20 Glyceryl Isostearate; PEG-30 Glyceryl Isostearate; PEG-60 Glyceryl Isostearate; PEG-12 Glyceryl Laurate; PEG-20 Glyceryl Laurate; PEG-23 Glyceryl Laurate; PEG-30 Glyceryl Laurate; PEG-10 Glyceryl Oleate; PEG-15 Glyceryl Oleate; PEG-30 Glyceryl Oleate; PEG-20 Glyceryl Ricinoleate; PEG-5 Glyceryl Sesquioleate; PEG-5 Glyceryl Stearate; PEG-10 Glyceryl Stearate; PEG-25 Glyceryl Stearate; PEG-30 Glyceryl Stearate; PEG-120 Glyceryl Stearate; PEG-200 Glyceryl Stearate; PEG-28 Glyceryl Tallowate; PEG-80 Glyceryl Tallowate; PEG-200 Glyceryl Tallowate; PEG-5 Glyceryl Triisostearate; PEG-5 Hydrogenated Castor Oil; PEG-7 Hydrogenated Castor Oil; PEG-16 Hydrogenated Castor Oil; PEG-20 Hydrogenated Castor Oil; PEG-25 Hydrogenate Castor Oil; PEG-30 Hydrogenate Castor Oil; PEG-35 Hydrogenate Castor Oil; PEG-40 Hydrogenate Castor Oil; PEG-45 Hydrogenate Castor Oil; PEG-50 Hydrogenate Castor Oil; PEG-54 Hydrogenate Castor Oil; PEG-55 Hydrogenate Castor Oil; PEG-60 Hydrogenate Castor Oil; PEG-80 Hydrogenate Castor Oil; PEG-100 Hydrogenate Castor Oil; PEG-200 Hydrogenate Castor Oil; PEG-40 Hydrogenated Castor Oil PCA Isosterate; PEG-5 Hydrogenated Corn Glycerides; and PEG-8 Hydrogenated Fish Glycerides; which are all available from known commercial sources

The solutions of the present invention may contain other additives including but not limited to buffers, tonicity agents, demulcents, wetting agents, preservatives, sequestering agents (chelating agents), surface active agents, and enzymes.

Other aspects of the claimed solutions include adding to the solution from 0.001 to 1 weight percent chelating agent (preferably disodium EDTA) and/or additional microbicide, (preferably 0.00001 to 0.1 or 0.0000 1 to 0.01) weight percent polyhexamethylene biquanide (PHMB0, Nalkyl-2-pyrrolidone, chlorhexidine, polyquaternium- 1, hexetidine, bronopol, alexidine, low concentrations of hydrogen peroxide, and ophthalmologically acceptable salts thereof

Ophthalmologically acceptable chelating agents useful in the present invention include amino carboxylic acid compounds or water-soluble salts thereof, including ethylenediaminetetraacetic acid, nitrilotriacetic acid, diethylenetriamine pentaacetic acid,

hydroxyethylethylenediaminetriacetic acid, 1,2-diaminocyclohexanetetraacetic acid, ethylene glycol bis (beta-aminoethyl ether) in N, N, N', N' tetraacetic acid (EGTA), aminodiacetic acid and hydroxyethylamino diacetic acid. These acids can be used in the form of their water soluble salts, particularly their alkali metal salts. Especially preferred chelating agents are the di-, tri- and tetra-sodium salts of ethylenediaminetetraacetic acid (EDTA), most preferably disodium EDTA (Disodium Edetate).

Other chelating agents such as citrates and polyphosphates can also be used in the present invention. The citrates which can be used in the present invention include citric acid and its mono-, di-, and tri-alkaline metal salts. The polyphosphates which can be used include pyrophosphates, triphosphates, tetraphosphates, trimetaphosphates, tetrametaphosphates, as well as more highly condensed phosphates in the form of the neutral or acidic alkali metal salts such as the sodium and potassium salts as well as the ammonium salt.

The pH of the solutions should be adjusted to be compatible with the eye and the contact lens, such as between 6.0 to 8.0, preferably between 6.8 to 7.8 or between 7.0 to 7.6. Significant deviations from neutral (pH 7.3) will cause changes in the physical parameters (i.e. diameter) in some contact lenses. Low pH (pH less than 5.5) can cause burning and stinging of the eyes, while very low or very high pH (less than 3.0 or greater than 10) can cause ocular damage.

The additional preservatives employed in the present invention are known, such as polyhexamethylene biguanide, N-alkyl-2-pyrrolidone, chlorhexidine, polyhexamethylenebiguanide, alexidine, polyquaternium- 1, hexetidine, bronopol and a very low concentration of hydrogen peroxide, e.g., 30 to 200 ppm.

The solutions of the invention are compatible with both rigid gas permeable and hydrophilic contact lenses during storage, cleaning, wetting, soaking, rinsing and disinfection.

A typical aqueous solution of the present invention may contain additional ingredients which would not affect the basic and novel characteristics of the active ingredients described earlier, such as tonicity agents, surfactants and viscosity inducing agents, which may aid in either the

lens cleaning or in providing lubrication to the eye. Suitable tonicity agents include sodium chloride, potassium chloride, glycerol or mixtures thereof The tonicity of the solution is typically adjusted to approximately 240-310 milliosmoles per kilogram solution (mOsm/kg) to render the solution compatible with ocular tissue and with hydrophilic contact lenses. In one embodiment, the solution contains 0.01 to 0.5 weight percent sodium chloride.

Suitable viscosity inducing agents can include lecithin or the cellulose derivatives such as hydroxymethylcellulose, hydroxypropylcellulose and methylcellulose in amounts similar to those for surfactants, above.

EXAMPLE 1

Hydrophilic contact lenses were placed flat onto glass slides and rinsed with water to remove any debris. These slides were placed in a petri dish and covered with a few drops of each of the test solutions previously prepared in either water, an aqueous isotonic sodium chloride solution, or an aqueous phosphate buffered solution made isotonic with sodium chloride and adjusted to pH 7.3. Each *petri* plate was covered and placed in a refrigerator overnight. The following day, the slides were removed and allowed to equilibrate to room temperature. The lenses were rinsed with water and the excess water was removed. One 5 uL drop of mineral oil stained with Oil Red O was placed onto one lens for each solution. After ten minutes, the lenses were observed for the ability of the oil drop to spread.

		Oil	Water
Additive	Solution Matrix	Dispersibility	Dispersibility
1% polyoxyl 40	water	4	5
hydrogenated castor oil			
(Cremophor RH 40)			
1% polyoxyl 40	buffer	5	5
hydrogenated castor oil	water		
(Cremophor RH 40)			
1% polyoxyl 40	sodium choride	2	5

hydrogenated castor oil	water		
(Cremophor RH 40)			Acceptable 1
1% polyoxyl 40	buffer	3	5
hydrogenated castor oil	sodium chloride		
(Cremophor RH 40)	water		
1% Polysorbate 80	sodium choride	4	5
(Tween 80)	water		
1% Poloxamine 1107	sodium choride	2	5
(Tetronic 1107)	water		
1% Poloxamer 407	sodium choride	2	5
(Pluronic F127)	water		
1% Polysorbate 80	buffer	3	5
(Tween 80)	sodium chloride		
	water		
1% Poloxamine 1107	buffer	1	5
(Tetronic 1107)	sodium chloride		
	water		
1% Poloxamer 407	buffer	1	5
(Pluronic F127)	sodium chloride		
	water		
Water	water	1	5

Key 1 non-spreading drop
2 poor spreading drop
3 moderate spreading drop
4 increased spreading drop

5 thin spreading film

The results demonstrates that exposure of the contact lens to the ethoxylated glyceride will generate a durable modified surface capable of allow the formation of a thin oil and aqueous film. This characteristic mimics mucin and is essential for the proper tear layer formation of

over the lens. A score of 3 or better is considered acceptable. This experiment also illustrates the synergistic improvement when the ethoxylated glyceride is exposed in the presence of a buffer. The inability of the Poloxamer and Poloxamine to allow the oil film to spread across the lens demonstrates that not all surface active agents will promote the spreading of a properly formed tear film over the contact lens surface.

EXAMPLE 2

Example of Protein Deposition Inhibition

Contact lenses were soaked and heated in test solutions to which a radio-labeled lysozyme was present in a known amount for a period of 12 hours at 37 degrees Celsius. The lenses were rinsed with distilled water in order to remove residual solution. The lenses were then assayed for protein deposition using a Beckman BioGamma 1 counter. Results were reported in ug/lens.

	Lens A ug/lens	Lens B ug/lens	Average ug/lens
Phosphate buffer control	1,043	865	954
Cremophor RH40 (1%) In Phosphate Buffer	15	23	19

Ethoxylated Castor Oil was a 1 percent w/v solution. The matrix control was phosphate buffer and sodium chloride. The polyoxyl 40 hydrogenated castor oil solution had lower protein binding than the control.

EXAMPLE 3

Example of Protein Deposition Inhibition

Isotonic aqueous phosphate buffered solutions were prepared and adjusted to pH 7.4. Contact lenses were soaked in 25 mL of the test solutions overnight. Afterwards, lysozyme was added to the tubes and warmed to 37 degrees Celsius for 12 hours. The lenses were rinsed with distilled water in order to remove residual solution. The lenses were assayed for protein deposition by the BCA method and detected on an HP PDA Spectrophotometer. Results were reported in ug/lens.

Solution	ug lysozyme per lens
Marketed Product Control	>18.3
(phosphate buffer, Poloxamer)	
Phosphate buffer control	>26.16
Cremophor RH40 (1%)	9.78
In Phosphate Buffer	

Ethoxylated Castor Oil was a 1 percent w/v solution. The matrix control was phosphate buffer and sodium chloride. The polyoxyl 40 hydrogenated castor oil solution had lower protein binding than the control.

EXAMPLE 4

An example of a preferred disinfecting formulation of the subject invention is provided below in Table I. This solution is prepared by weighing out the necessary amount of the tricine, creatine, choline chloride, sodium chloride and edetate disodium into a vessel containing approximately 90% of the water volume. After each of the ingredients has dissolved, the pH is adjusted to 7.3 with either 1 N sodium hydroxide or 1 N hydrochloric acid. Following this, the polyhexamethylene biguanide is added and the solution is brought to final volume with purified water. The final product has the composition shown in the Table below.

Constituent		Weight / Volume
Polyhexamethylenebiguanide HCl	20% w/w solution available under the mark Cosmocil CQ, from Avecia	0.0001%
Tricine	Spectrum	1.0%
Creatine	Spectrum	0.25%
Choline Chloride	Amersco	0.5%
Edetate Disodium	Spectrum	0.055%
Polyoxyl 40 Hydrogenated Castor Oil	Cremophor RH 40 from BASF Co.	0.1%
Sodium Chloride	Fisher Scientific	As required for

		tonicity adjustment
		300 mOsm
Hydrochloride Acid, 1N	VWR	as required for pH
		adjustment to 7.3
Sodium Hydroxide, 1N	Mallinckrodt	as required for pH
		adjustment to 7.3
Purified Water		Balance to 100%

This solution may be used to rinse, clean, and store contact lenses on a daily basis.

EXAMPLE 5

An example of a preferred formulation for a contact lens vial storage of the subject invention is provided below in Table I. This solution is prepared by weighing out the necessary amount of the sodium borate, boric acid, and sodium chloride into a vessel containing approximately 90% of the water volume. After each of the ingredients has dissolved, the pH is adjusted to 7.3 with either 1 N sodium hydroxide or 1 N hydrochloric acid. The final product had the composition shown in Table I below.

Constituent		Weight / Volume
Sodium Borate	Spectrum	1.0%
Boric Acid	Spectrum	0.25%
Polyoxyl 40 Hydrogenated Castor Oil	Cremophor RH40 from BASF Co.	0.1%
Sodium Chloride	Fisher Scientific	As required for tonicity adjustment 300 mOsm
Hydrochloride Acid, 1N	VWR	as required for pH adjustment to 7.3
Sodium Hydroxide, 1N	Mallinckrodt	as required for pH adjustment to 7.3
Purified Water		Balance to 100%

Example 6

The following are useful disinfecting solutions within the scope of the present invention that may be used for all purpose disinfecting solutions. They are made according to generally acceptable procedures except that the ethoxylated glycerides must be first be dissolved in warm water prior to the addition of the other components.

Constituent	Supplier	% Weight/	Amount
		Volume	
Purified water		to 80%	40 mL
Tricine	Spectrum	1.0%	0.500 g
Carnitine	Spectrum	0.25%	0.125 g
Betaine HCl	Spectrum	0.1%	0.050 g
Choline Chloride	Amresco	0.5%	0.250 g
Inositol	Spectrum	0.1%	0.050 g
Edetate Disodium	Spectrum	0.055%	0.0275 g
Polyoxyl 40 Hydrogenated	Cremophor RH 40	0.1%	0.5 mL of 10%
Castor Oil	from BASF Co.		
Hydrochloride Acid, 1N		as required for pH	as required for pH
		adjustment to 7.3	adjustment to 7.3
Sodium Hydroxide, 1N		as required for pH	as required for pH
		adjustment to 7.3	adjustment to 7.3
Purified Water		to 98%	Dilute to 49 mL
Sodium Chloride	Fisher	As required for	As required for
		tonicity adjustment	tonicity adjustment
		300 mOsm	300 mOsm
Polyhexamethylene-	20% w/w solution	0.0001%	50 uL of 0.1%
biguanide HCl	available under		
	the mark Cosmocil		
	CQ from Avecia		
Purified Water		Balance to 100%	Dilute to 50 mL

Example 7

The following are formulations within the scope of the invention of formulations intended to be used as lens-vial solutions that are used to store lenses prior to their use. These solutions have the effect of treating the contact lens in the solution and rendering the lens more comfortable in use.

Constituent	Supplier	% Weight/	Amount
		Volume	
Purified water		to 80%	40 mL
Tricine	Spectrum	1.0%	0.500 g
Carnitine	Spectrum	0.25%	0.125 g
Inositol	Spectrum	0.1%	0.050 g
Hydrochloride Acid, 1N		as required for pH	as required for pH
		adjustment to 7.3	adjustment to 7.3
Sodium Hydroxide, 1N		as required for pH	as required for pH
		adjustment to 7.3	adjustment to 7.3
Polyoxyl 40 Hydrogenated	Cremophor RH 40	0.1%	0.5 mL of 10%
Castor Oil	from BASF Co.		
Purified Water		to 98%	Dilute to 49 mL
Sodium Chloride	Fisher	As required for	As required for
		tonicity adjustment	tonicity adjustment
		300 mOsm	300 mOsm
Purified Water		to 100%	Dilute to 50 mL

What is claimed is:

- An ophthalmic contact lens solution comprising:
 0.001 to 10 percent by weight ethoxylated glyceride;
 0.001 to 2 weight percent of a physiologically acceptable buffer adjusted so the pH of solution is between 6.5 and 7.8 and the balance water.
- An ophthalmic contact lens solution comprising:
 0.001 to 10 percent by weight ethoxylated glyceride;
 0.001to 2 weight percent of a physiologically acceptable tonicity agent adjusted so the solution is isotonic between 200 and 400 mOsm
- An ophthalmic solution comprising;
 0.001 to 10 percent by weight ethoxylated glyceride;
 0.00001 to 0.1 weight percent of a preservative agent.
- 4. The solution of claim 1 which further comprises 0.01 to 2 weight percent of a physiologically acceptable tonicity agent adjusted so the solution is isotonic between 200 and 400 mOsm
- 5. The solution of claim 4 that further comprises 0.00001 to 0.1 weight percent of a preservative.
- 6. The solution of claim 1 wherein the ethoxylated glyceride is chosen from the group of compounds consisting of Polyoxyl 40 hydrogenated castor oil (Cremophor RH 40), polyoxyl 60 hydrogenated castor oil (Cremophor RH 60), PEG-30 Castor Oil (Incrocas 30), PEG-35 Castor Oil (Cremophor EL, Incrocas 35), or PEG-40 Castor Oil (Cremophor EL, Incrocas), Cremophor EL ®, Emulphor EL ®, glycerol polyethyleneglycol riciinoleate, gycerol polyethyleneglycol oxystearate, polyethoxylated hydrogenated castor oil, or polyethoxylated vegetable oil.

- 7. The solution of claim 1 wherein the buffer is selected from the group consisting of organic amines, organic carboxylic acids, amphoterics, phosphates, or borates.
- 8. Method for rendering a contact lens wettable by contacting the surface of said lens with an aqueous solution comprising from .001 to about 10 precent by weight of an ethoxylated glyceride.
- 9. The method of claim 8 wherein the the ethoxylated glyceride is polyoxyl 40 hydrogenated castor oil.
- 10. The method of claim 7 wherein said ethoxylated glyceride is polyoxyl 60 hydrogenated castor oil.
- 11. The method of claim 7 wherein said ethoxylated glyceride is polyoxyl 40 hydrogenated castor oil.
- 12. The method of claim 7 wherein said ethoxylated glyceride is polyoxyl 35 castor oil.
- 13. The method of claim 7 wherein the aqueous solution further comprises the buffer bis(2-hydroxyethyl)iminotris(hydroxymethyl)methane (Bis–Tris) and its salts.
- 14. The method of claim 7 wherein the aqueous solution further comprises the 1,2-bis[tris(hydroxymethyl)-methylamino) propane (Bis-Tris Propane) and its salts.
- 15. The method of claim 7 wherein the aqueous solution further comprises the N-tris(hydroxymethyl) methyl glycine (Tricine) and its salts.
- The method of claim 7 wherein the aqueous solution further comprises the N,N-bis(2-hydroxyethyl)-glycine (Bicine) and its salts.
- 17. The method of claim 7 wherein the aqueous solution further comprises the betaine and its salts.
- 18. The method of claim 7 wherein the aqueous solution further comprises the buffer phosphate and its salts
- 19. The method of claim 7 wherein the aqueous solution further comprises the buffer is borate and its salts
- 20. The method of claim 7 wherein the aqueous solution further comprises the is citrate and

its salts

- 21. The method of claim 7 wherein the aqueous solution further comprises is TRIS and its salts
- 22. The method of claim 7 wherein the aqueous solution further comprises the buffer is 2-amino-2-methyl-1,3-propanediol and its salts
- 23. The method of claim 7 wherein the aqueous solution further comprises the buffer is triisopropanolamine and its salts
- 24. The method of claim 7 wherein the aqueous solution further comprises the buffer is carnitine and its salts
- 25. The method of claim 7 wherein the aqueous solution further comprises the buffer is dimethyl glutamate and its salts
- 26. The method of claim 7 wherein the aqueous solution further comprises the buffer is creatine and its salts
- 27. The method of claim 7 wherein the aqueous solution further comprises the buffer is diethanolamine and its salts
- 28. The method of claim 7 wherein the aqueous solution further comprises the buffer is diisopropylamine and its salts
- 29. The method of claim 7 wherein the aqueous solution further comprises the buffer is triethanolamine and its salts
- 30. The method of claim 7 wherein the aqueous solution further comprises the buffer is triethylamine and its salts
- The method of claim 7 wherein the aqueous solution further comprises the buffer is dimethyl aspartic acid and its salts
- 32. The method of claim 7 wherein the aqueous solution further comprises the buffer is imidazole and its salts
- 33. The method of claim 7 wherein the aqueous solution further comprises the buffer is histidine and its salts
- 34. The method of claim 7 wherein the aqueous solution further comprises the buffer is methyl aspartate and its salts
- The method of claim 7 wherein the aqueous solution further comprises the buffer is Tris(hydroxymethyl)aminomethane (Tromethamine, TRIS) and its salts

35. A contact lens product comprising:

A contact lens:

A sealable container; and

An effective amount of an ophthalmic lens solution comprising:

0.001 to 10 percent by weight ethoxylated glyceride;

0.01 to 2 weight percent of a physiologically acceptable buffer adjusted so the pH of solution is between 6.5 and 7.8 and the balance water.

- 37. The method of claim 7 wherein the buffer is glycine and its salts
- 38. The method of claim 7 wherein the buffer is lysine and its salts
- 39. The method of claim 7 wherein the buffer is histidine and its salts.,

Abstract

An ophthalmic solution comprising a polyethoxylated glyceride in the range of 0.001 to about 10 percent by weight and a buffer agent. These solutions impart surprising comfort and wearability to contact lenses. At the same time the solutions provide good preservative capacity and do not increase protein deposit.

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DECLARATION — Utility or Design Patent Application

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Direct all correspondence to: Customer Number or Bar Code Label OR Correspondence address below						
Name Chris Blank						
Address 4 Bicentennial Square						
Address SUITE 2B						
city CONCORD	· · · · · · · · · · · · · · · · · · ·	***************************************		State	NA	ZIP 03301
country USA	1	elephone	6032	275	248	Fax 603 227 5263
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NAME OF SOLE OR FIRST INVENTOR:						
Given Name (first and middle [if any]) FRANCIS X Family Name or Surname Smith						
Inventor's France X for Date 11-3-00						
Residence: City Sale M			State /	SH	country SA	Citizenship VSA
Mailing Address 22 Fox Run						
Mailing Address						
city Salem	State) H		ZIP		Country BA
NAME OF SECOND INVENTOR:						
Given Name (first and middle [if any]) John Randall Family Name or Surname						
Inventor's Signature Date						
Residence: City			State		Country	Citizenship
Mailing Address						
Mailing Address		·				
City State ZIP						Country
Additional inventors are being named on thesupplemental Additional Inventor(s) sheet(s) PTO/SB/02A attached hereto.						